

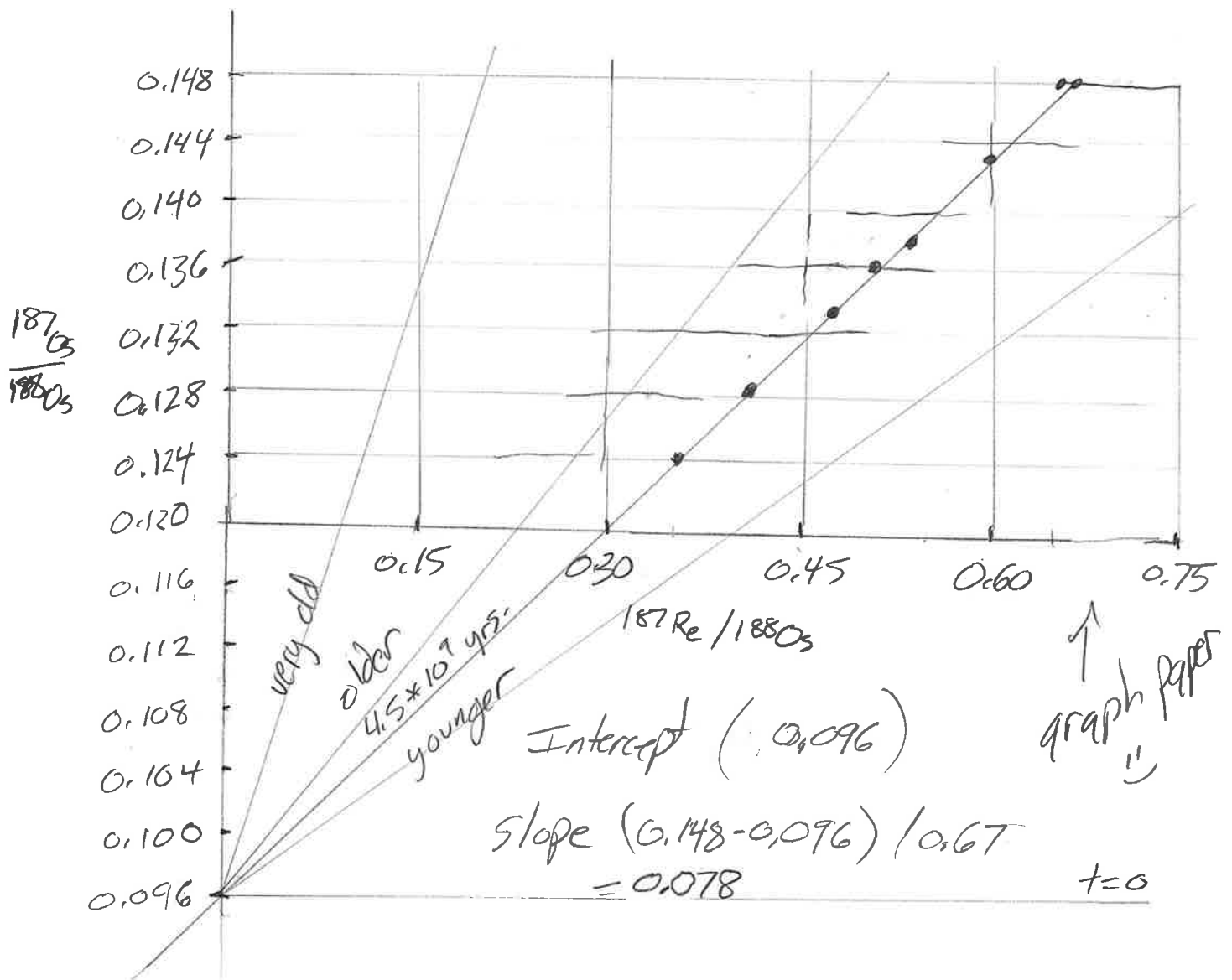
ASTR 430 Homework #5

Solution Set

Hamilton

Problem 11-10

- a) Age Dating Meteorites using
 b) $^{187}\text{Re} \rightarrow ^{187}\text{Os}$ w/ $t_{1/2} = 4.16 \times 10^{10}$ yrs.
 c)



Now, see Fig 11.18 and use the equation given there for slope

$$\text{slope} = e^{(t-t_0)/t_m} - 1$$

where $\Delta T = t - t_0 = \text{elapsed time}$

$$\text{Eq. 11.5} \rightarrow t_{1/2} = \ln 2 t_m = 4.16 \times 10^{10} \text{ yrs.}$$

$$\Rightarrow t_m = 6.00 \times 10^{10} \text{ yrs.}$$

$$0.078 = e^{\Delta T / 6.00 \times 10^{10}} - 1$$

$$\frac{\Delta T}{6.00 \times 10^{10}} = \ln(0.078 + 1) = 0.075$$

$$\Rightarrow \Delta T = 4.5 \times 10^9 \text{ yrs.}$$

Initial Abundance of
(y-intercept)

$$\frac{187_{\text{Os}}}{188_{\text{Os}}} = 0.096$$

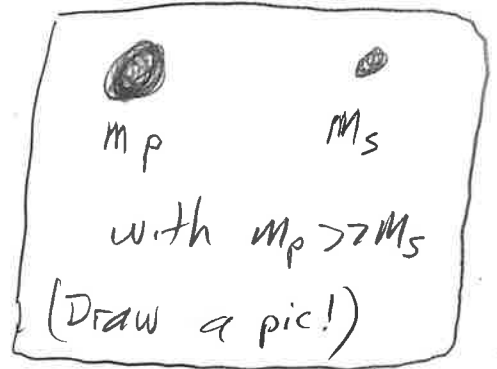
Problem 13-1

Hill sphere vs. Roche Limit

These concepts are related. The Hill sphere (radius R_H) is

the region around m_S in which satellites can stably orbit.

The Roche Limit (distance r_R) is the distance from m_P at which m_S can be tidally disrupted. In each case, forces from m_P are disrupting the m_S system. The force doing the disrupting in each case is the tidal force (= M_P gravity + centrifugal force),



a) Derive 13.7 from 13.6

$$\left(\frac{a}{R_p} \right) = 3^{1/3} \left(\frac{\rho_p}{\rho_s} \right)^{1/3}$$

$$\frac{GM_s}{R_s^2} = \frac{3GM_p R_s}{a^3}$$

solve 13.6 for a^3 and use
 $M_p = \frac{4}{3} \pi R_p^3 \rho_p$ and $M_s = \frac{4}{3} \pi R_s^3 \rho_s$

$$\Rightarrow a^3 = 3 R_s^3 \left(\frac{M_p}{M_s} \right) = 3 R_s^3 \frac{R_p^3 \rho_p}{R_s^3 \rho_s}$$

$$\left(\frac{a}{R_p} \right) = 3^{1/3} \left(\frac{\rho_p}{\rho_s} \right)^{1/3}$$

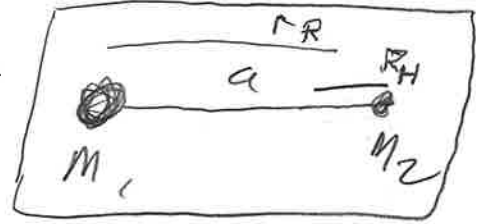
13.7

b) Derive 13.7 from 2.28

$$R_H = \left(\frac{M_2}{3(M_1 + M_2)} \right)^{1/3} a$$

Note that the
 a 's are formally
 different $a = r$ from a) is distance from
 M_1 to the limit. a from b) is the

distance between the two masses.
 But not an issue since $a \gg R_H$.
 Now we know we need
 to get rid of masses - that's
 easy so start there.



$$\frac{M_2}{M_1 + M_2} = \frac{\rho_2 R_2^3}{\rho_1 R_1^3 + \rho_2 R_2^3} \approx \frac{\rho_2 R_2^3}{\rho_1 R_1^3}$$

where we use $M_1 \gg M_2$

$$\text{so } \left(\frac{a}{R_H}\right)^3 = 3 \left(\frac{\rho_1}{\rho_2}\right) \frac{R_1^3}{R_2^3}$$

$$\underbrace{\left(\frac{a}{R_H}\right) \left(\frac{R_2}{R_1}\right)}_{\uparrow} = 3^{1/3} \left(\frac{\rho_1}{\rho_2}\right)^{1/3} \leftarrow \text{RHS of 13.7}$$

$\rho_1 = \rho_p ; \rho_2 = \rho_s$

This equals
 a/R_p if $R_2 = R_H$ \leftarrow Is this
 justified?

$R_2 = R_H$ basically says that if the satellite fills its Hill sphere, then it will be disrupted because the edge of the satellite reaches the Roche limit. ✓

Done! 😊

Problem 15-3 Interstellar Gas Cloud

a) On a circular orbit at 1 AU, an H_2 molecule has Energy

$$E = \frac{1}{2} m_H v^2 - \frac{GMm_H}{r} = -\frac{GMm_H}{2a}$$

where M = the Sun's mass

m_H = mass of H_2 molecule

v = velocity

r = distance from the Sun

a = semimajor axis

So the total energy is

$$E = \frac{-(6.67 \times 10^{-11}) (2 \times 10^{30} \text{ kg}) (2.167 \times 10^{-27} \text{ kg})}{2 (1.496 \times 10^{11} \text{ m})}$$

$$= -1.49 \times 10^{-18} \text{ J}$$

The velocity of the H_2 molecule along its circular orbit will be:

$$v = \sqrt{GM \left(\frac{2}{r} - \frac{1}{a} \right)} = \sqrt{\frac{GM}{r}} = 2.99 \times 10^4 \text{ m/s} \\ = 29.9 \text{ km/s}$$

b) This is clearly a conservation of Energy problem. We balance the energy of the H_2 molecule before it drops down to 1 AU against the energy afterwards. In symbols:

$$\textcircled{1} \quad E_{\text{before}} = E_{\text{orbit}} + E_{\text{heat}}$$

We can find E_{before} from

$$E = \frac{1}{2} m_H v^2 - \frac{GMm_H}{r} = 0 \quad \text{when } r = \infty, v = 0$$

So $\textcircled{1}$ becomes

$$0 = -1.49 \times 10^{-18} \text{ J} + E_{\text{heat}}$$

$$E_{\text{heat}} = 1.49 \times 10^{-18} \text{ J}$$

From Physics, $E_{\text{heat}} = \frac{3}{2} kT$ where

k is Boltzmann's constant $k = 1.38 \times 10^{-23} \text{ J/K}$

$$\text{So } T = \frac{2}{3k} \cdot (1.49 \times 10^{-18} \text{ J})$$

Hot!

$$T = 72,000 \text{ K}$$

This is the primary source of heat for the solar nebula, the disk of mostly Hydrogen and Helium that circled the protosun. Most of the energy released by freefall escapes as radiation, but a large amount is left over for the disk. Since more energy is released as gas falls further in toward the central mass concentration, we expect that the solar nebula's temperature will increase as we approach the protostar.

Later, fusion starts in the protostar, and another source of heat is tapped.

Problem 15-12

Find smallest body melted by decay of ^{26}Al

Heating rate initially: $10^{-3} \frac{\text{J}}{\text{m}^3 \text{s}}$

$$\lambda_m^{-1} = 8 \times 10^{-7} \text{ yr}^{-1}$$

$$c_p = 700 \text{ J/kg K}$$

$$\text{diffusivity } 10^{-6} \text{ m}^2/\text{s}$$

$$T_{\text{melt}} = 1800 \text{ K}$$

Following the hint, we calculate time-scales for 1) heat to diffuse and 2) to raise T to T_{melt}

1) Diffusivity - no help from the book :-

wikipedia: Diffusion Equation

$$\text{is } \frac{\partial T}{\partial t} = \alpha \nabla^2 T \quad \text{w/ } \alpha = \text{diffusivity}$$

Looks like this is a rough estimation problem, why? We ignore shape and composition of the asteroid, its initial temperature, etc.

So diffusion timescale is, dimensionally,

$$\text{time} = \frac{(\text{length})^2}{\alpha}$$

The heating timescale is just

$$\begin{aligned} \tau_m &= \frac{1}{8 \times 10^{-7} \text{ yr.}} = 1.25 \times 10^6 \text{ yr.} \\ &= 3.9 \times 10^{13} \text{ s} \end{aligned}$$

Equate!

$$\frac{r^2}{10^6} = 3.9 \times 10^{13} \Rightarrow r = \sqrt{3.9 \times 10^7}$$

$$r = 6240 \text{ m} = 6.2 \text{ km}$$

so for Asteroids $> 6.2 \text{ km}$, the heat builds up faster than it diffuses away.

Done? Not quite. We have to check whether enough heat is generated to melt the body.

$$\text{so } H \approx \dot{H} \tau_m = C_p m \cdot \Delta T$$

estimate of
total heating

Temperature
rise

check if $\Delta T > 1800 \text{ K}$

$$\Delta T = \frac{\dot{H} \tau_m}{C_p m} ; \dot{H} = 10^{-3} \frac{\text{J}}{\text{m}^3 \text{s}} (\text{Volume})$$

$$M = \rho(\text{Volume})$$

↖ estimate at 3000 kg/m^3

$$\Rightarrow dT = \frac{10^{-3} \text{ J/m}^3 (\text{Volume}) 3.9 \times 10^{13} \text{ s}}{(700 \text{ J/kg K}) 3000 \text{ kg/m}^3 (\text{Volume})}$$

$$= \frac{3.9 \times 10^{10}}{2.1 \times 10^6} = \boxed{1.9 \times 10^4 \text{ K}}$$

Hot!

plenty of energy in decay
of ^{26}Al to melt rock + iron.